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Preliminary communication

CYCLOPENTADIENYL-RUTHENIUM AND -OSMIUM CHEMISTRY. SYNTHESES AND DEPROTONATION OF SOME VINYLIDENE-RUTHENIUM COMPLEXES.

Michael I. Bruce and Robert C. Wallis

Department of Physical and Inorganic Chemistry, University of Adelaide, . Adelaide, South Australia, 5001.

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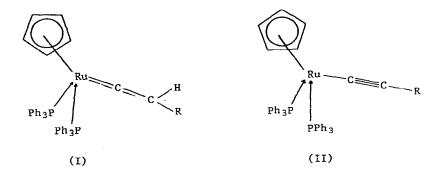
Summary. Reactions between 1-alkynes and RuCl(PPh₃)₂(n-C₅H₅) in the presence of NH₄PF₆ afford the cationic vinylidene complexes [Ru(C:CHR)(PPh₃)₂(n-C₅H₅)]PF₆; these are readily deprotonated by base to give the η^1 -alkynyl derivatives Ru(C=CR)(PPh₃)₂(n-C₅H₅). The latter may be protonated to reform the monosubstituted vinylidene complexes.

Recent publications have described the isolation and characterisation of dicyanovinylidene complexes of molybdenum¹, and of phenylvinylidene complexes of manganese^{2,3} and rhenium⁴. The latter were obtained in low yield from $M(CO)_3(n-C_5H_5)$ (M = Mn or Re) and phenylacetylene in tetrahydrofuran. Others have implicated similar complexes of platinum, PtC⁺=CiIR, as intermediates in the formation of the methoxycarbene complexes trans-PtMe[C(OMe)CH₂R]L₂ (L = PHe₂Ph, AsMe₃) from reactions between 1-alkynes and trans-PtClMeL₂ in methanol in the presence of AgPF₆,⁵ and of ruthenium in the dimerisation of t-butylacetylene to 1,4-di-t-butylbuta-1,2,3-triene by RuH₂(CO)(PPh₃)₃.⁶ The reaction between phenylacetylene and trans-FeCl₂(depe)₂ [depe = Et₂P(CH₂)₂PEt₂] affords a complex thought to contain the C:CHPh ligand on the basis of IR, ¹H NMR and Mössbauer spectrometry.⁷ Herein we describe a simple, high yield synthesis of some cationic vinylidene complexes of ruthenium, their characterisation, and conversion into J-alkynyl derivatives. The reactions of RuCl(PPh₃)₂(n-C₅H₅)⁸ with 1-alkynes in methanol, in the presence of NH₄PF₆ (or NaBPh₄) give pale red-purple compounds of stoichiometry [Ru(C₂HR)(PPh₃)₂(n-C₅H₅)]PF₆ (or BPh₄) (R = Me, Ph, C₆F₅ or CO₂Me, for example). The IR and NMR data [v(CC) 1640 cm⁻¹; CH at τ 5.1t J_{HP} < 1Hz); RuCCPh at 350 p.p.m.] are consistent with their formulation as the n¹-vinylidene complexes (I). In particular, the triplet found for the metal-bonded carbon in the ¹³C NMR spectra occurs in the region (250-360 p.p.m.) usually associated with carbene complexes,⁹ in contrast to the value of 112.8 p.p.m. found for Pt(n²-C₂Me₂)(PPh₃)₂.¹⁰ In the present case, the strong deshielding observed can be rationalised by the efficient back-bonding from the electron-rich metal centre into the strongly π-accepting vinylidene ligand. The vinylidene formulation is also supported by the large gem H-Me coupling (ca. 7Hz) found for (I; R = Me), obtained from prop-1-yne.

Deprotonation of the vinylidene complexes occurs on treatment with bases such as hydroxide, carbonate, or methylide, with formation of the corresponding σ -acetylide. The same reaction occurs on treatment of the cationic complexes with alumina, when the initially dark red-purple solutions rapidly change to the yellow of the neutral Ru(CECR)(PPh₃)₂(n-C₅H₅) (II) complexes. These are readily isolated as yellow crystalline solids, and are characterised by their spectroscopic properties [v(CC) *ca*. 2070 cm⁻¹, ¹H NMR, no resonance assignable to CH; ¹³C NMR, \equiv C at *ca*. 230 p.p.m.]. Compounds of this type have been reported previously, e.g. (II, R = CO₂Me) from the reaction between RuMe(PPh₃)₂(n-C₅H₅) and HC₂CO₂Me,¹¹ and as copper(I) chloride adducts, from RuCl(PPh₃)₂(n-C₅H₅) and CuC₂R.¹²

The stability of the cationic vinylidene complexes suggested that an alternative route to them might be via the protonation of the neutral ethynyl derivatives. Addition of HBF_4 or HPF_6 to $Ru(C_2Ph)(PPh_3)_2(\eta-C_5H_5)$ afforded directly at room temperature the phenylvinylidene complex (I), and other substituted ethynyl complexes react similarly. This reaction appears to have some general applicability to the formation of vinylidene complexes

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from the corresponding σ -acetylides. However, $trans-Pt(C_2CF_3)_2(PEt_3)_2$ does not react with HPF₆ over 1 h at room temperature, while alkoxycarbene complexes are obtained from other platinum(II) acetylides and strong acids in alcohol solvents.¹³

There are several points of interest raised by this work, and we are currently investigating the factors which determine whether n^2 -alkyne complexes, or the isomeric n^1 -vinylidene complexes, are obtained from reactions between transition metal complexes and 1-alkynes.

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